# Versatile 2-aminothiazoles, building blocks for highly functionalised heterocycles 

Gerd Kaupp [a]*, Fathy A. Amer [b], Mohamed Abbas Metwally [b] and Ehab Abdel-latif [b]

[a] University of Oldenburg, Faculty 5, Organic Chemistry I, P.O. Box 2503, 26111 Oldenburg, Germany;
Fax: +49 4417983409
[b] University of Mansoura, Faculty of Sciences, Egypt; Fax: +20 502246781.
Received April 10, 2003
The reactions of quantitatively available 4-phenyl- and 4-(4-antipyrinyl)-2-aminothiazole ["4-antipyrinyl-" is used as a short-term for "4-(1,5-dimethyl-3-oxo-2-phenyl-2,3-dihydro-1-H-pyrazol-4yl)-"] with chloroacetyl chloride, acetic anhydride, ethyl cyanoacetate and carbon disulphide are reported. The products are transformed further by Knoevenagel condensations and coupling reactions with aromatic diazonium salts. The latter occur both at the thiazole ring and at the active methylene sites. The tautomerism of these products is studied on the basis of density functional theory calculations at the B3LYP/6-31G* level.
J. Heterocyclic Chem., 40, 963 (2003).

Introduction.
The 2-aminothiazole building block is of widespread use in chemistry [1], medicine [2-5] and pharmacology (for example antibacterial [6] or anti-inflammatory [7] applications). A large number of 2-aminothiazoles have been substituted with different groups for pharmaceutical purposes [8]. These syntheses are largely facilitated by the ease of access of 2-aminothiazoles when using the new waste-free solid-state synthesis techniques from thioureas and $\alpha$-halogenoketones [9]. We explored the synthetic capabilities in view of varied and diverse reactions of 4-phenyland 4-(4-antipyrinyl)-2-aminothiazol in order to point out the versatility even in the presence of complicated functionality. Actually two pharmacophores are linked together in the latter system. The primary amino group of the 2 -aminothiazoles can be substituted in various ways.

Furthermore, secondary substitutions, cyclizations, Knoevenagel condensations and azo couplings, the latter to the thiazole ring and to the introduced cyanoacetamide substituent, will be performed in order to provide numerous highly functionalised heterocycles. The chromophores that were introduced might be useful as labels or for dying purposes of the model compounds.
Results.

## 1. 2-Aminothiazoles.

4-Phenyl-2-aminothiazole (1) and related compounds have been prepared by waste-free solid-state reaction of $\alpha$ halogenoketones and thioureas with $100 \%$ yield [9]. The same technique could be applied to 4-(chloroacetyl)antipyrine (2) and thiourea (3). The product 4 (that had been previously obtained in $92 \%$ yield by conventional

Scheme 1


Scheme 2

synthesis [8]) could now be obtained with $100 \%$ yield from stoichiometric reactions by ball-milling followed by washings with $\mathrm{Na}_{2} \mathrm{CO}_{3}$ solutions (Scheme 1). This remarkable 3-cascade reaction of substitution, cyclization and dehydration proceeds again [9] in the solid state. The solid-state mechanism of such unusual processes has been amply demonstrated by nanoscopic techniques [9-10].
Numerous 2-aminothiazoles such as compound 1 have been reacted with aliphatic or aromatic aldehydes to form the corresponding imines (Schiff bases) and their antimicrobial and antifungal activities tested [11]. Numerous further reactions of the primary amino group of the reagents $\mathbf{1}$ and 4 appear useful.
2. 2-Thiazolylamides and 2-Thiazolyl-cyanoacetamides.

2-( $N$-acetylamino)-4-phenyl-thiazole and many substituted derivatives have been described [12]. Similarly, treatment of compound 4 with acetic anhydride at $65^{\circ}$ yielded the mono-acetylation product 2-( $N$-acetylamino)-4-(4-antipyrinyl)-thiazole (5) (Scheme 2).

Reaction of compound 4 with an equimolar amount of chloroacetylchloride in DMF containing some drops of triethylamine afforded the corresponding 4-(4-antipyrinyl)-2-( $\alpha$-chloroacetamido)-thiazole (6) (Scheme 2). The spectral data of the amides 5 and $\mathbf{6}$ were consistent with their structures. Furthermore, compound 6 reacted with 2 -mer-capto-benzothiazole, -benzoxazol and -benzimidazol in ethanol solution containing a few drops of triethylamine to yield the thioethers 7a-c. The structures of 7a-c were established by spectral data (see Experimental). Interestingly, products of the type 7 (e.g. phenyl instead of the 4 -antipyrinyl substituent) exhibited antimicrobial activity [13] and the present compounds may be tested accordingly.
The 2-aminothiazoles $\mathbf{1}$ and $\mathbf{4}$ reacted with ethyl cyanoacetate and sodium ethoxide in ethanol solution and afforded the corresponding cyanoacetamide derivatives $\mathbf{8}$ with an active methylene group for further derivatization (see Section 3). The molecular structures of compounds $\mathbf{8 a}$ [14] and $\mathbf{8 b}$ [15] are clearly indicated by the spectral data.

Scheme 3

3. N -(2-Thiazolyl)-2-propenamides.

The reactivity of the compounds $\mathbf{8}$ toward Knoevenagel condensation was tested with 4-methylbenzaldehyde and
the corresponding derivatives $\mathbf{9 a}, \mathbf{b}$ were isolated (Scheme 4). The molecular structure of these products was confirmed on the basis of their spectral data.

Scheme 4


## 4. Reactions of $\mathbf{1}$ and $\mathbf{4}$ with Carbon Disulphide.

The base promoted nucleophilic addition of the 2aminothiazoles $\mathbf{1}$ and $\mathbf{4}$ to an equimolar amount of carbon disulphide in DMF containing potassium hydroxide afforded the potassium sulphide salts $\mathbf{1 0}$ that were not isolated. Subsequent treatment of the salts $\mathbf{1 0}$ with an equimolar amount or an excess of methyl iodide and base furnished the mono- or di-methylated products 11b and 12b, respectively (Scheme 5). The compounds 11a [16] and 12a [17] are already known, the molecular structures of 11b and 12b were secured by spectral data.

Dimethyl dithiocarbonimidates are useful precursors for the synthesis of many important heterocyclic compounds $e . g$. arylamino- or heteroarylamino-1,3-heteroazoles [18], $ß$-lactams [19], 3-heteroaryl-tetrahydroquina-zolin-2,4-diones [20], 8-arylamino- or heteroarylaminopurines [21] and they serve as dienes in Diels-Alder reactions [22]. We tried the reaction of $\mathbf{1 2 b}$ with 2 -aminothio-phenol in refluxing $\mathrm{N}, \mathrm{N}$-dimethylformamide (DMF) in the presence of one equivalent of sodium hydroxide and obtained a single product that was identified to be compound $\mathbf{1 4 b}$, based on analytical and spectral data. Heterocyclization of $\mathbf{1 0}$ was achieved by reaction with chloroacetic acid (after neutralization with sodium carbonate). The 3-substituted thiazolidinone-2thione derivatives $\mathbf{1 3 a}, \mathbf{b}$ were obtained. The molecular structure of $\mathbf{1 3}$ was confirmed by analytical and spectral data. Thus, interesting heterocycles are easily accessible by these routes and extensions for further reactions can be predicted.

## 5. Coupling of 8, $\mathbf{9}$ and $\mathbf{1 3}$ with Aryl Diazonium Salts.

The compounds $\mathbf{8}$ have two active sites for electrophilic substitution reactions with aromatic diazonium salts. The methylene group in compound 8a proved to be more reactive toward the azo coupling reaction [23] with diazonium salts than the free C-5 position in the thiazole ring. Thus, an equimolar amount of aryl diazonium chlorides at $0-5^{\circ} \mathrm{C}$ reacted with compound 8a to yield the corresponding monohydrazone derivatives $\mathbf{1 5}$. If two moles of aryl

Scheme 5

diazonium chlorides and pyridine were applied at $0-5{ }^{\circ} \mathrm{C}$ the coupling at the thiazole ring occurred as well and the products 16 were obtained (Scheme 6). Compound 16 was also obtained by azo-coupling of the monohydrazone derivatives 15. The molecular structure of the compounds $\mathbf{1 5}$ and $\mathbf{1 6}$ derives from the analytical and spectral data. The tautomeric structure of product 15 was elucidated by density functional calculations at the B3LYP/6-31G* level. The most stable tautomer found exhibits the hydrazone structure 15 with one intramolecular hydrogen bond. The alternative hydrazone structure $\mathbf{1 5}^{\prime}$ was only 1.4 kcal $\mathrm{mol}^{-1}$ worse. Therefore, both should be in equilibrium. However, the best possible azo structure was found 13.7 kcal $\mathrm{mol}^{-1}$ higher in energy and that excludes their significant contribution to the equilibrium.

The azo couplings of compound $\mathbf{8 b}$ could not avoid the reaction at the thiazole ring. If one mole or two moles of the diazonium salt were applied only the corre-
drawing electron density from the 2 -amide function. The couplings at C-5 to give $\mathbf{1 6}$ and $\mathbf{1 7}$ are unprecedented. Contrary to these results, no azo coupling at the thiazole ring of $\mathbf{8 b}$ had previously been reported when heterocyclic diazonium salts reacted only at the active methylene group [15].


Interestingly, the compounds $\mathbf{1 6}$ and $\mathbf{1 7}$ have both a hydrazone and an azo moiety in the same molecule. Density functional model calculations at the B3LYP/6$31 G^{*}$ level (structure 16 but with the phenyl group and the cyanohydrazone carbon both replaced by H and $\mathrm{Ar}=\mathrm{Ph}$ ) excluded a double hydrazone structure for $\mathbf{1 6}$ or $\mathbf{1 7}$ by the estimated energy difference of $11.4 \mathrm{kcal} \mathrm{mol}^{-1}$.

The methylene group in the compounds $\mathbf{1 3}$ was not active enough for the azo coupling. Thus, only electrophilic coupling at the thiazole site was observed with equimolar or excessive amounts of aryl diazonium chlorides at $0-5^{\circ}$ (Scheme 8). The molecular structures of the monoazo derivatives $\mathbf{1 8}$ and $\mathbf{1 9}$ were established by analytical and spectroscopic data.

The azo coupling of compound $\mathbf{9}$ without a competing active methylene group occurred easily, as expected, to afford the products 20 and 21 (Scheme 9). The structure of the highly functionalized dyes $\mathbf{2 0}$ and $\mathbf{2 1}$ was assigned on the basis of spectral data.

Scheme 6

sponding bis-azo derivatives $\mathbf{1 7}$ could be isolated (Scheme 7). This fact is explained by the 4 -antipyrinyl group which donates electron density to the thiazole ring (the ${ }^{13} \mathrm{C}$ nmr shifts of the $5-\mathrm{C}$ of $\mathbf{8 a}$ and $\mathbf{8 b}$ are 106.5 and 102.9 ppm , respectively). It appears that initial coupling to the imidazole ring enhances the reactivity of the remote methylene group by additionally with-

Scheme 8


Scheme 9


Discussion.
A large number of substituted 2-aminothiazoles has been synthesized on useful paths. These highly functionalised derivatives may be of interest to pharmacy and for dyeing purposes, yet to be explored. Very high and complicated functionality is not only achieved by coupling to the antipyrine pharmacophore but also by further functionalization of the substituents at the 2-amino group. Thirty two 2-thiazolyl-amides, -cyanoacetamides, -sulfanylacetamides, -propenamides, -arylhydrazonoacetamides, -dithiocarbamates, -dithiocarbonimidates, -rhodanines, -aminobenzothiazoles and 5-arylazo-thiazolylamides were prepared and characterized. All of these products exhibit unusually high functionality with varied groups, which demonstrates the stability of the synthetic pathways that are opened by the waste-free access of the 2-aminothiazole building blocks. The azo couplings at the cyanoacetamide groups and at the thiazole ring are particularly versatile and show marked variations in reactivity. Interesting conjugated systems arise in compounds 15-21. While the absorption maxima in the uv/visible spectra do not reach 500 nm , the dyeing properties may be favourable due to the high concentration of heteroatoms. These features will be explored in due time.

## EXPERIMENTAL

General Aspects.
Melting points were determined with a Gallenkamp melting point apparatus (capillary method) and are uncorrected. Elemental analyses (C, H, N; Table 1) were carried out at the Microanalytical Unit of the Faculty of Science, Mansoura University, Egypt. Infrared spectra (ir) were recorded with a Perkin-Elmer 1720-X FT-IR spectrometer using potassium bromide pellets (not all frequencies are reported), uv/visible spectra with a Perkin-Elmer Lambda 551 S spectrometer. All nmr spectra were acquired using a Bruker WP 300 spectrometer at $300 \mathrm{MHz}\left({ }^{1} \mathrm{H}\right)$ or 75.5 MHz ${ }^{13} \mathrm{C}$; in broad band mode). Deuteriochloroform/dimethylsulfox-ide- $\mathrm{d}_{6}\left(\mathrm{CDCl}_{3} / \mathrm{DMSO}-\mathrm{d}_{6}\right)$ mixtures contained up to $20 \%$ DMSO$\mathrm{d}_{6}$. All $\delta$-values are given in ppm and refer to the internal standard tetramethylsilane (TMS), $\psi \mathrm{d}$ in ${ }^{1} \mathrm{H} \mathrm{nmr}$ stands for an apparent doublet of a higher order spin system. J-Values are given in Hertz $(\mathrm{Hz})$. Mass spectra were obtained at a Finnigan MAT 212 instrument by electron impact at 70 eV . For known compounds spectral data were restricted to previously not available data. The ball-mill for 2 mmol runs was a Retsch MM 2000 swing mill with a 10 ml
stainless steel double-walled beaker with fittings for a heating bath. Water of the appropriate temperature was circulated for heating or cooling. Two stainless steel balls with 12 mm diameter were used. Ball-milling was performed at $20-25 \mathrm{~Hz}$ frequency. Progress and completeness of the solid-state reactions was checked by ir spectroscopy in potassium bromide. The product yields were detected by weight, product purity by mp , thin layer chromatography (tlc, on Merck Silica Gel $60 \mathrm{~F}_{254}$ plates) and ${ }^{1} \mathrm{H}$ nmr spectroscopy. The short-term "4-antipyrinyl-" is used for the 4-(1,5-dimethyl-3-oxo-2-phenyl-2,3-dihydro-1- H -pyrazol-4-yl)substituent in order to improve the readability. B3LYP (basis set 6-31G*) density functional theory calculations with full geometry optimization were performed with the program TITAN, version 1.01, of Wavefunction, Inc., Irvine, USA.

## 2-Amino-4-(4-antipyrinyl)-thiazole (4).

A mixture of 4-(chloroacetyl)-antipyrine (2) [8] (497 mg, 2.00 $\mathrm{mmol})$ and thiourea (3) ( $152 \mathrm{mg}, 2.00 \mathrm{mmol}$ ) was ball-milled at $70^{\circ}$ for 60 minutes. After drying at 0.01 bar at $80^{\circ}$ a quantitative yield of $\mathbf{4} \cdot \mathrm{HBr}(615 \mathrm{mg}, 100 \%)$ was obtained. The free base 4 was obtained by washing the fine powder of the hydrobromide with $5 \%$ aqueous $\mathrm{Na}_{2} \mathrm{CO}_{3}$ solution followed by water and drying at $80^{\circ}$ in a vacuum. The yield was $530 \mathrm{mg}(100 \%) ; \mathrm{mp} 235-236^{\circ}$ (Lit. [8] mp 235 $)$; ir (KBr): 3327, 3151, 2956, 1636, $1593 \mathrm{~cm}^{-1}$; ${ }^{1} \mathrm{H} \mathrm{nmr}\left(\mathrm{DMSO}_{\mathrm{d}}^{6}\right.$ ): $\delta 2.70(\mathrm{~s}, 3 \mathrm{H}), 3.15(\mathrm{~s}, 3 \mathrm{H}), 6.55(\mathrm{~s}, 2 \mathrm{H})$, 7.15 (s, 1H, thiazol proton), $7.25-7.50(\mathrm{~m}, 5 \mathrm{H}$, phenyl protons); ${ }^{13} \mathrm{C} \mathrm{nmr}\left(\right.$ DMSO-d $\left.\mathrm{d}_{6}\right): \delta 10.8,33.9,100.4,102.7,122.5$ (2C), $124.8,127.3$ (2C), 133.6, 141.0, 150.6, 162.36, 165.6.
$N$-[4-(4-Antipyrinyl)-thiazol-2-yl]-acetamide (5).
A mixture of compound $\mathbf{4}(1.43 \mathrm{~g}, 5.0 \mathrm{mmol})$ and acetic anhydride ( 3.0 ml ) was heated in an oil bath at $60-65^{\circ}$ for 1 hour. The reaction mixture was allowed to cool at room temperature and then recrystallized from ethanol to afford $1.35 \mathrm{~g}(82 \%)$ colorless crystals, mp $227^{\circ}$; ir (KBr): 3185 (NH), 3094, 2964, 1689 (amide $\mathrm{C}=\mathrm{O}$ ), 1631 (pyrazole $\mathrm{C}=\mathrm{O}$ ), 1542, 1491, 1371, $1278 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ $\mathrm{nmr}\left(\mathrm{CDCl}_{3}\right): \delta 2.10(\mathrm{~s}, 3 \mathrm{H}), 2.60(\mathrm{~s}, 3 \mathrm{H}), 3.10(\mathrm{~s}, 3 \mathrm{H}), 7.25-7.45$ $(\mathrm{m}, 5 \mathrm{H}$, phenyl protons), $7.70(\mathrm{~s}$, thiazol proton), $10.30(\mathrm{~s}, 1 \mathrm{H}$, $\mathrm{NH}) ;{ }^{13} \mathrm{C} \mathrm{nmr}\left(\mathrm{CDCl}_{3}\right): \delta$ 13.1, 23.7, 36.0, 105.2, 110.1, 125.7 (2C), 128.0, 129.9 (2C), 135.5, 142.1, 152.48, 157.9, 164.9, $168.8 ; \mathrm{ms}: \mathrm{m} / \mathrm{z}(\%) 328\left(\mathrm{M}^{+}, 100\right), 313,286,271,243,226,194$, 166, 138, 93, 77, 56; hrms: calcd. for $\mathrm{C}_{16} \mathrm{H}_{16} \mathrm{~N}_{4} \mathrm{O}_{2} \mathrm{~S}\left(\mathrm{M}^{+}\right)$ 328.0994, found 328.0983.
$N$-[4-(4-Antipyrinyl)-thiazol-2-yl]- $\alpha$-chloroacetamide (6).
Chloroacetyl chloride ( $1.2 \mathrm{ml}, 15 \mathrm{mmol}$ ) was dropwise added with stirring to a solution of compound $4(2.86 \mathrm{~g}, 10.0 \mathrm{mmol})$ in DMF ( 15 ml ) containing triethyl amine ( 0.5 ml ) at room temperature. Stirring was continued for 4 hours and the reaction mixture was poured to ice cooled water. The precipitate was collected by filtration, dried and recrystallized from ethanol to obtain 2.61 g ( $72 \%$ ) colorless crystals, $\mathrm{mp} 208^{\circ}$; ir ( KBr ): 3282, 1700, 1615, 1590, 1549, 1487, 1455, 1410, 1322, 1260, 1155, 1082, 753 $\mathrm{cm}^{-1} ;{ }^{1} \mathrm{H} \mathrm{nmr}\left(\mathrm{CDCl}_{3}\right): \delta 2.70(\mathrm{~s}, 3 \mathrm{H}), 3.20(\mathrm{~s}, 3 \mathrm{H}), 4.25(\mathrm{~s}, 2 \mathrm{H})$, $7.25-7.50(\mathrm{~m}, 5 \mathrm{H}$, phenyl protons), $7.85(\mathrm{~s}, 1 \mathrm{H}$, thiazol proton), $9.80(\mathrm{~s}, 1 \mathrm{H}, \mathrm{NH}) ;{ }^{13} \mathrm{C} \mathrm{nmr}\left(\mathrm{CDCl}_{3}\right): \delta 12.53,35.40,42.06$, $104.42,110.11,124.60$ (2C), 127.04, 129.21 (2C), 134.91, $142.12,151.95,155.50,163.60,164.21 ; \mathrm{ms}: \mathrm{m} / \mathrm{z}(\%) 364\left(\mathrm{M}^{+}\right)$, $362\left(\mathrm{M}^{+}, 100\right), 328,313,286,243,213,194,138,93,77,56 ;$ hrms: calcd. for $\mathrm{C}_{16} \mathrm{H}_{15} \mathrm{ClN}_{4} \mathrm{O}_{2} \mathrm{~S} 362.0604$ ( $\mathrm{M}^{+}$), found 362.0605.

General Procedure for the Synthesis of 2-(Azol-sulfanyl)-N-[4-(4-antipyrinyl)-thiazol-2-yl]-acetamide Derivatives (7a-c).
A mixture of 4-(4-antipyrinyl)-2-( $\alpha$-chloroacetamido)thiazole $6(1.08 \mathrm{~g}, 3.0 \mathrm{mmol})$ and the corresponding 2-mercaptobenzazole ( 3.0 mmol ) was refluxed in ethanol ( 50 ml ) containing 5 drops of triethyl amine for 4 hours. The precipitate upon cooling was collected by filtration and crystallized from ethanol.
2-(Benzothiazol-2-ylsulfanyl)- $N$-[4-(4-antipyrinyl)-thiazol-2-yl]acetamide (7a).

This compound was obtained as colorless crystals, $\mathrm{mp} 120^{\circ}$; yield: 920 mg ( $62 \%$ ); ir ( KBr ): 3174, 3057, 2963, 1695, 1641, 1591, 1562, 1488, 1455, 1424, 1299, 1238, 1120, 1080, 1006, $796,755 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H} \mathrm{nmr}\left(\mathrm{CDCl}_{3}\right): \delta 2.60(\mathrm{~s}, 3 \mathrm{H}), 3.10(\mathrm{~s}, 3 \mathrm{H})$, $4.15(\mathrm{~s}, 2 \mathrm{H}), 7.20-7.50(\mathrm{~m}, 7 \mathrm{H}$, phenyl protons), $7.75(\psi \mathrm{~d}, 1 \mathrm{H}$, phenyl proton), 7.85 ( $\mathrm{s}, 1 \mathrm{H}$, thiazol proton), 8.00 ( $\psi \mathrm{d}, 1 \mathrm{H}$, phenyl proton), 11.90 (br s, $1 \mathrm{H}, \mathrm{NH}$ ); ${ }^{13} \mathrm{C} \mathrm{nmr}\left(\mathrm{CDCl}_{3}\right): \delta 12.4,35.4$, $36.1,104.5,109.4,121.2,121.4,124.4$ (2C), 125.1, 125.6, 126.3, 126.8, 129.1 (2C), 134.9, 135.4, 142.0, 151.9, 156.2, 164.2, $165.8,167.1 ; \mathrm{ms}: \mathrm{m} / \mathrm{z}(\%) 493$ (43, M ${ }^{+}$), 419, 327, 313 (100), 286, 243, 208, 180, 108, 77, 56; hrms: calcd. for $\mathrm{C}_{23} \mathrm{H}_{19} \mathrm{~N}_{5} \mathrm{O}_{2} \mathrm{~S}_{3}$ $\left(\mathrm{M}^{+}\right) 493.0701$, found 493.0698.

2-(Benzoxazol-2-ylsulfanyl)-N-[4-(4-antipyrinyl)-thiazol-2-yl]acetamide (7b).
This compound was obtained as colorless crystals, $\mathrm{mp} 182^{\circ}$; yield: 960 mg ( $67 \%$ ); ir ( KBr ): 3194, 3100, 2969, 1702, 1645 , 1592, 1562, 1510, 1455, 1413, 1370, 1318, 1264, 1239, 1216, 1134, 1098, 1049, $755 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H} \mathrm{nmr}\left(\mathrm{CDCl}_{3}\right): \delta 2.60(\mathrm{~s}, 3 \mathrm{H})$, $3.10(\mathrm{~s}, 3 \mathrm{H}), 4.10(\mathrm{~s}, 2 \mathrm{H}), 7.20-7.50(\mathrm{~m}, 8 \mathrm{H}$, phenyl protons), $7.60(\psi \mathrm{~d}, 1 \mathrm{H}$, phenyl proton), $7.75(\mathrm{~s}, 1 \mathrm{H}$, thiazol proton), 11.60 (br s, $1 \mathrm{H}, \mathrm{NH}$ ); ${ }^{13} \mathrm{C} \mathrm{nmr}\left(\mathrm{CDCl}_{3}\right): \delta 12.3,34.8,35.2,104.3$, $109.4,110.3,118.1,124.6$ (3C), 124.7, 127.0, 129.1 (2C), 134.8, $140.8,141.8,151.8,152.2,156.1,164.1,164.9,165.3 ; \mathrm{ms}: \mathrm{m} / \mathrm{z}$ (\%) $477\left(\mathrm{M}^{+}\right), 403,313$ (100), 286, 243, 192, 164, 151, 105, 78.
2-(1H-Benzimidazol-2-ylsulfanyl)- N -[4-(4-antipyrinyl)-thiazol-2-yl]-acetamide (7c).
This compound was obtained as colorless crystals, mp $161^{\circ}$; yield: 900 mg ( $63 \%$ ); ir (KBr): 3385, 3171, 3082, 2991, 1683, 1627, 1590, 1562, 1497, 1441, 1416, 1348, 1310, 1268, 1133, 972, 836, 794, $746 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H} \mathrm{nmr}\left(\mathrm{CDCl}_{3} / \mathrm{DMSO}-\mathrm{d}_{6}\right): \delta 2.75(\mathrm{~s}$, 3 H ), $3.20(\mathrm{~s}, 3 \mathrm{H}), 4.25(\mathrm{~s}, 2 \mathrm{H}), 7.10-7.50(\mathrm{~m}, 9 \mathrm{H}$, phenyl protons), 7.65 (s, 1H, thiazol proton), 12.40 (br s, $2 \mathrm{H}, 2 \mathrm{NH}$ ); ms: m/z (\%) $476\left(\mathrm{M}^{+}\right), 442,402,388,313,286(100), 243,194,118,91,77$.

General Procedure for the Synthesis of N -(Thiazol-2-yl)-cyanoacetamides (8).

The 2-aminothiazole $\mathbf{1}$ or $\mathbf{4}(20 \mathrm{mmol})$ and ethyl cyanoacetate ( $2.13 \mathrm{ml}, 20 \mathrm{mmol}$ ) were added to a solution of sodium ethoxide (from $0.46 \mathrm{~g} \mathrm{Na}, 20 \mathrm{mmol}$ ) in absolute ethanol $(30 \mathrm{ml})$. The reaction mixture was refluxed for 6 hours with stirring and the solvent was evaporated under reduced pressure. The residue was dissolved in water and neutralized by 1 N HCl . The separated solid was collected by filteration and recrystallized from ethanol.

## $N$-(4-Phenyl-thiazol-2-yl)-cyanoacetamide (8a).

This compound was obtained as colorless crystals, mp $195^{\circ}$ (Lit. [14] mp $192^{\circ}$ ); yield: $3.30 \mathrm{~g}(68 \%)$; ir (KBr): 3371, 3171, 3053, 2948, 2263 (CN), 1700, 1662, 1571, 1485, 1445, 1389, 1338, $1283,1178 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H} \mathrm{nmr}\left(\mathrm{CDCl}_{3}\right): \delta 3.95(\mathrm{~s}, 2 \mathrm{H}), 7.25-7.40(\mathrm{~m}$,
$3 \mathrm{H}, 2$ phenyl protons and thiazol proton), 7.85 ( $\psi \mathrm{d}, 2 \mathrm{H}$, phenyl protons), $12.50(\mathrm{~s}, 1 \mathrm{H}, \mathrm{NH}) ;{ }^{13} \mathrm{C} \mathrm{nmr}\left(\mathrm{CDCl}_{3}\right): \delta 24.3,106.5$, 113.0, 124.2 (2C), 126.2, 127.0 (2C), 132.7, 147.9, 155.9, 159.8.
$N$-[4-(4-Antipyrinyl)-thiazol-2-yl]-cyanoacetamide (8b).
This compound was obtained as colorless crystals, mp 219$220^{\circ}$ (Lit. [15] mp 225-226 ${ }^{\circ}$ ); yield: 5.43 g ( $77 \%$ ); ir (KBr): 3391, 3166, 3062, 2969, 2260 (CN), 1697 (C=O), 1631, 1562, 1489, 1407, 1340, 1271, 1155, 1049, 942, 882, 837, 750, $695 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ $\mathrm{nmr}: \delta 2.70(\mathrm{~s}, 3 \mathrm{H}), 3.20(\mathrm{~s}, 3 \mathrm{H}), 3.80(\mathrm{~s}, 2 \mathrm{H}), 7.30-7.50(\mathrm{~m}, 5 \mathrm{H}$, phenyl protons), $7.70(\mathrm{~s}, 1 \mathrm{H}$, thiazol proton), 12.20 (br s, 1 H , NH); ${ }^{13} \mathrm{C} \mathrm{nmr}\left(\mathrm{DMSO}_{\mathrm{d}}^{6}\right): ~ \delta 11.4,24.9,34.3,102.9,107.8$, 113.3, 123.7 (2C), 126.0, 128.1 (2C), 133.9, 141.2, 150.8, 155.3, 159.7, 163.0; ms: m/z (\%) 353 ( $\mathrm{M}^{+}, 100$ ), 313, 286, 243, 194, 143, 93, 77, 56; hrms: calcd. for $\mathrm{C}_{17} \mathrm{H}_{15} \mathrm{~N}_{5} \mathrm{O}_{2} \mathrm{~S}\left(\mathrm{M}^{+}\right) 353.0946$, found 353.0946.

General Procedure for the Synthesis of 2-Cyano-N-[4-(hetero)aryl-2-thiazolyl)-3-(4-tolyl)-2-propenamides (9).

A mixture of $\mathbf{8}(10 \mathrm{mmol})$ and $p$-methylbenzaldehyde $(1.2 \mathrm{ml}$, $10 \mathrm{mmol})$ in 20 ml ethanol and two drops of piperidine was refluxed for 2 hours. The solid products that separated on cooling were collected by filtration, dried and crystallized from ethanol.
2-Cyano-N-(4-phenyl-2-thiazolyl)-3-(4-tolyl)-2-propenamide (9a).

This compound was obtained as light yellow crystals, mp 202웅 yield: 2.35 g ( $68 \%$ ); ir ( KBr ): 3378, 3115, $2213(\mathrm{CN})$, 1682, 1596, 1542, 1442, 1327, 1283, 1177, 816, $728 \mathrm{~cm}^{-1},{ }^{1} \mathrm{H}$ $\mathrm{nmr}\left(\mathrm{CDCl}_{3}\right): \delta 2.40(\mathrm{~s}, 3 \mathrm{H}), 7.20(\mathrm{~s}, 1 \mathrm{H}), 7.30-7.40(\mathrm{~m}, 5 \mathrm{H}$, phenyl protons), 7.85 ( $\psi \mathrm{d}, 2 \mathrm{H}$, phenyl protons), $7.90(\psi \mathrm{~d}, 2 \mathrm{H}$, phenyl protons), $8.40(\mathrm{~s}, 1 \mathrm{H}$, thiazol proton), $9.70(\mathrm{~s}, 1 \mathrm{H}, \mathrm{NH})$; ${ }^{13} \mathrm{C} \mathrm{nmr}\left(\mathrm{CDCl}_{3}\right): \delta 21.9,100.8,108.7,116.2,126.1$ (2C), 128.2, 128.8 (2C), 128.8, 130.2 (2C), 131.3 (2C), 134.0, 145.3, 150.5, 155.0, 156.8, 158.6; hrms: calcd. for $\mathrm{C}_{20} \mathrm{H}_{15} \mathrm{~N}_{3} \mathrm{OS}\left(\mathrm{M}^{+}\right)$ 345.0936, found 328.0935.

2-Cyano- $N$-[4-(4-antipyrinyl)-thiazol-2-yl]-3-(4-tolyl)-2-propenamide (9b).

This compound was obtained as brown shiny crystals, mp $232^{\circ}$; yield: $3.32 \mathrm{~g}(73 \%)$; ir ( KBr ): 3403, 3127, 2972, 2212 (CN), 1677, 1651, 1610, 1586, 1537, 1489, 1446, 1279, 1179, $1146 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H} \mathrm{nmr}\left(\mathrm{CDCl}_{3}\right): \delta 2.40(\mathrm{~s}, 3 \mathrm{H}), 2.70(\mathrm{~s}, 3 \mathrm{H}), 3.20(\mathrm{~s}$, 3 H ), 7.30-7.50 ( $\mathrm{m}, 7 \mathrm{H}$, phenyl protons), $7.80(\mathrm{~s}, 1 \mathrm{H}$, thiazol proton), $7.90(\psi \mathrm{~d}, 2 \mathrm{H}$, phenyl protons), $8.45(\mathrm{~s}, 1 \mathrm{H}), 9.75(\mathrm{~s}, 1 \mathrm{H}$, $\mathrm{NH}) ;{ }^{13} \mathrm{C} \mathrm{nmr}\left(\mathrm{CDCl}_{3}\right): \delta 12.7,21.89,35.3,101.3,103.2,109.8$, 116.5, 124.6 (2C), 127.0, 129.0, 129.2 (2C), 130.2 (2C), 131.3 (2C), 134.8, 141.6, 145.1, 152.0, 154.8, 156.5, 158.9, 164.1; ms: $\mathrm{m} / \mathrm{z}(\%) 455\left(\mathrm{M}^{+}, 100\right), 387,353,313,243,170,115,77$; hrms: calcd. for $\mathrm{C}_{25} \mathrm{H}_{21} \mathrm{~N}_{5} \mathrm{O}_{2} \mathrm{~S}\left(\mathrm{M}^{+}\right) 455.1416$, found 455.1426 .
Methyl- $N$-[4-(4-antipyrinyl)-thiazol-2-yl]-dithiocarbamate (11b).
To a well stirred solution of $4(2.86 \mathrm{~g}, 10.0 \mathrm{mmol})$ in DMF ( 20 ml ) in an ice-bath were added dropwise and successively aqueous potassium hydroxide ( 0.56 g in 5 ml water), carbon disulphide $(0.60 \mathrm{ml}, 10 \mathrm{mmol})$ and methyl iodide ( $0.65 \mathrm{ml}, 10 \mathrm{mmol}$ ). Stirring was continued for 3 hours and the mixture was poured into water. The solid thus obtained was collected by filtration, washed with water and recrystallized from ethanol to yield $1.96 \mathrm{~g}(52 \%)$ colorless crystals, $\mathrm{mp} \mathrm{198}{ }^{\circ}$; ir ( KBr ): 3441, 3082, 2908, 1626, 1615, 1592, 1479, 1397, 1331, 1288, 1234, 1216, 1163, 1088, 1001, 949,
$776,693 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H} \mathrm{nmr}\left(\mathrm{CDCl}_{3} / \mathrm{CF}_{3} \mathrm{COOD}\right): \delta 2.60(\mathrm{~s}, 3 \mathrm{H}), 2.80$ ( $\mathrm{s}, 3 \mathrm{H}$ ), $3.50(\mathrm{~s}, 3 \mathrm{H}), 7.00(\mathrm{~s}, 1 \mathrm{H}$, thiazol proton), $7.35(\mathrm{~s}, 2 \mathrm{H}$, phenyl protons), 7.60 (s, 3 H , phenyl protons); ${ }^{13} \mathrm{C} \mathrm{nmr}$ ( $\left.\mathrm{CDCl}_{3} / \mathrm{CF}_{3} \mathrm{COOD}\right): \delta 11.5,19.0,33.3,94.2,104.6,128.3$ (2C), $129.9,130.3,130.8$ (2C), 132.3, 146.7, 159.1, 162.1, 197.4; ms: $\mathrm{m} / \mathrm{z}(\%) 376\left(\mathrm{M}^{+}\right), 353,328$ (100), 253, 194, 165, 135, 76; hrms: calcd. for $\mathrm{C}_{16} \mathrm{H}_{16} \mathrm{~N}_{4} \mathrm{OS}_{3}\left(\mathrm{M}^{+}\right)$376.0486, found 376.0483.

Dimethyl- N -[4-(4-antipyrinyl)-thiazol-2-yl]-dithiocarbonimidate (12b).

To a well stirred solution of $4(2.86 \mathrm{~g}, 10 \mathrm{mmol})$ in DMF ( 20 ml ) in an ice-bath were dropwise and successively added aqueous potassium hydroxide ( 0.56 g in 5 ml water), carbon disulphide $(0.60 \mathrm{ml} ; 10 \mathrm{mmol})$, aqueous potassium hydroxide $(0.56 \mathrm{~g}$ in 5 ml water) and methyl iodide ( $1.3 \mathrm{ml} ; 21 \mathrm{mmol}$ ). Stirring was continued for 2-3 hours and the mixture was poured into water. The solid that separated was collected by filtration, washed with water and recrystallized from ethanol to yield $1.87 \mathrm{~g}(48 \%)$ colorless crystals, mp $166^{\circ}$; ir (KBr): 3108, 2956, 1651, 1593, 1510, 1482, 1456, 1423, 1395, 1345, 1312, 1240, 1192, 1145, 1065, 1034, $977,939,922,840,787 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H} \mathrm{nmr}\left(\mathrm{CDCl}_{3}\right): \delta 2.60(\mathrm{~s}$, $6 \mathrm{H}), 2.80(\mathrm{~s}, 3 \mathrm{H}), 3.15(\mathrm{~s}, 3 \mathrm{H}), 7.25-7.45(\mathrm{~m}, 5 \mathrm{H}$, phenyl protons), $8.00\left(\mathrm{~s}, 1 \mathrm{H}\right.$, thiazol proton); ${ }^{13} \mathrm{C} \mathrm{nmr}\left(\mathrm{CDCl}_{3}\right): \delta 12.7$, $15.9,35.5,104.8,112.7,124.3$ (2C), 126.7, 129.1 (2C), 135.1, $144.1,152.4,164.4,167.0,169.9 ; \mathrm{ms}: \mathrm{m} / \mathrm{z}(\%) 390\left(\mathrm{M}^{+}, 100\right)$, 343, 329, 317, 296, 236, 171, 110, 77, 56; hrms: calcd. for $\mathrm{C}_{17} \mathrm{H}_{18} \mathrm{~N}_{4} \mathrm{OS}_{3}\left(\mathrm{M}^{+}\right) 390.0643$, found 390.0647 .

General Procedure for the Synthesis of 3-(Thiazol-2-yl)-rhodanines (13).
The reaction mixture of the 2-aminothiazole derivative $\mathbf{1}$ or $\mathbf{4}$ $(10 \mathrm{mmol})$ and $\mathrm{KOH}(0.56 \mathrm{~g}$ in 4 ml water) and carbon disulphide $(0.60 \mathrm{ml}, 10 \mathrm{mmol})$ was stirred in DMF $(20 \mathrm{ml})$ at $0-5^{\circ}$ till a clear solution was obtained (about 2 hours). Simultaneously, chloroacetic acid ( $945 \mathrm{mg}, 10 \mathrm{mmol}$ that was neutralized by $\mathrm{K}_{2} \mathrm{CO}_{3}$ in 10 ml water) was added dropwise. The mixture was poured into water and acidified by 1 MHCl . The precipitate was collected by filtration, washed with water and recrystallized from DMF with addition of water.

## 3-(4-Phenyl-thiazol-2-yl)-rhodanine (13a).

This compound was obtained as colorless crystals, $\mathrm{mp} 202^{\circ}$; yield: 1.87 g ( $64 \%$ ); ir (KBr): 3115, 2945, 1682, 1594, 1568, 1518, 1452, 1408, 1325, 1238, 1175, $1013 \mathrm{~cm}^{-1}$; ${ }^{1} \mathrm{H} \mathrm{nmr}\left(\mathrm{CF}_{3} \mathrm{COOD}\right): \delta$ $4.25(\mathrm{~s}, 2 \mathrm{H}), 7.15(\mathrm{~s}, 1 \mathrm{H}$, thiazol proton), 7.50-7.65 (m, 5 H , phenyl protons); ${ }^{13} \mathrm{C} \mathrm{nmr}\left(\mathrm{CF}_{3} \mathrm{COOD}\right): ~ \delta 37.4,106.2,126.6$ (2C), 126.8, 129.9 (2C), 131.4, 141.4, 162.9, 174.1, 195.2; ms: m/z (\%) 292 $\left(\mathrm{M}^{+}\right), 264,250,218$ (100), 203, 186, 176, 134; hrms: calcd. for $\mathrm{C}_{12} \mathrm{H}_{8} \mathrm{~N}_{2} \mathrm{OS}_{3}\left(\mathrm{M}^{+}\right)$291.9799, found 291.9797.

3-[4-(4-Antipyrinyl)-thiazol-2-yl]-rhodanine (13b):
This compound was obtained as colorless crystals, $\mathrm{mp} 198^{\circ}$; yield: 2.21 g ( $55 \%$ ); ir ( KBr ): 3131, 2932, 1722, 1616, 1573, 1480, 1397, 1304, 1214, 987, $776 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H} \mathrm{nmr}\left(\mathrm{CF}_{3} \mathrm{COOD}\right): \delta$ $2.65(\mathrm{~s}, 3 \mathrm{H}), 3.55(\mathrm{~s}, 3 \mathrm{H}), 4.30(\mathrm{~s}, 2 \mathrm{H}), 7.15(\mathrm{~s}, 1 \mathrm{H}$, thiazol proton), $7.35-7.70\left(\mathrm{~m}, 5 \mathrm{H}\right.$, phenyl protons); ${ }^{13} \mathrm{C} \mathrm{nmr}\left(\mathrm{CF}_{3} \mathrm{COOD}\right): ~ \delta$ $11.5,33.3,37.5,94.3,105.4,128.5$ (2C), 129.9, 131.0 (2C), 132.7, 146.9, 159.0, 160.7, 162.6, 173.5, 194.1; ms: m/z (\%) 402 $\left(\mathrm{M}^{+}\right), 353,290,260,265,235$ (100), 105, 78, 63, 44; hrms: calcd. for $\mathrm{C}_{17} \mathrm{H}_{14} \mathrm{~N}_{4} \mathrm{O}_{2} \mathrm{~S}_{3}\left(\mathrm{M}^{+}\right) 402.0279$, found 402.0280 .

2-[4-(4-Antipyrinyl)-thiazol-2-yl]-aminobenzothiazole (14b).
A solution of 2-aminothiophenol ( $250 \mathrm{mg}, 2.0 \mathrm{mmol}$ ) in DMF $(10 \mathrm{ml})$ was heated with aqueous sodium hydroxide $(0.40 \mathrm{~g}$ in 3 ml water) and the mixture was stirred at room temperature for 30 minutes. A solution of $\mathbf{1 2 b}(780 \mathrm{mg}, 2.0 \mathrm{mmol})$ in DMF was then added dropwise and the reaction mixture heated under reflux for 4 hours. After cooling, the mixture was poured into water and neutralized with 1 M HCl . The precipitate thus obtained was collected by filtration, washed with water, dried and recrystallized from an ethanol-DMF mixture (1:1) to yield 470 mg ( $56 \%$ ) colorless crystals, mp $231^{\circ}$; ir (KBr): 3196, 3048, 2945, 1637, 1618, 1592, 1521, 1476, 1452, 1301, 1278, 1211, 1066, 1019, 912, 875, $753,725,700,640 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H} \mathrm{nmr}\left(\mathrm{CDCl}_{3} / \mathrm{CF}_{3} \mathrm{COOD}\right): \delta 2.65$ (s, $3 \mathrm{H}), 3.70(\mathrm{~s}, 3 \mathrm{H}), 7.40(\psi \mathrm{~d}, 2 \mathrm{H}$, phenyl protons), $7.50(\mathrm{~s}, 1 \mathrm{H}$, thiazol proton), 7.60-7.75 (m, 6H, phenyl protons), $7.85(\psi \mathrm{~d}, 1 \mathrm{H}$, phenyl proton); ${ }^{13} \mathrm{C} \mathrm{nmr}\left(\mathrm{CDCl}_{3} / \mathrm{CF}_{3} \mathrm{COOD}\right): ~ \delta 11.4,33.4,96.3$, 115.0, 123.1, 124.3, 126.7, 128.7 (2C), 129.5, 129.9, 131.1 (2C), $132.2,133.1,135.6,146.9,157.8,162.4,166.2,167.0 ; \mathrm{ms}: \mathrm{m} / \mathrm{z}$ (\%) $419\left(\mathrm{M}^{+}, 100\right), 387,335,321,286,243,215,176,135,108$, 91, 77; hrms: calcd. for $\mathrm{C}_{21} \mathrm{H}_{17} \mathrm{~N}_{5} \mathrm{OS}_{2}\left(\mathrm{M}^{+}\right) 419.0875$, found 419.0871.

General Procedure for the Synthesis of $N$-(4-Phenyl-thiazol-2-yl)-2-arylhydrazono-2-cyanoacetamides (15).

A solution of sodium nitrite ( $0.70 \mathrm{~g}, 10.0 \mathrm{mmol}$ in 10 ml water $)$ was gradually added to a well cooled solution of the aromatic amine ( 10.0 mmol ) in concentrated $\mathrm{HCl}(3 \mathrm{~mL})$. The diazonium salt solution was added with constant stirring to a cold solution of compound $\mathbf{8}(10.0 \mathrm{mmol})$ in ethanol ( 50 ml ) and sodium acetate $(4.0 \mathrm{~g})$. The reaction mixture was allowed to rest at $0^{\circ}$ for 2 hours and the solid was collected by filtration. The monoarylazo derivatives $\mathbf{1 5}$ thus obtained, were dried and recrystallized from an ethanol-DMF mixture (1:1).
$N$-(4-Phenyl-thiazol-2-yl)-2-phenylhydrazono-2-cyanoacetamide (15a).

This compound was obtained as yellow crystals, mp $247^{\circ}$; yield: $2.64 \mathrm{~g}(76 \%)$; uv/visible $\left(\mathrm{CH}_{3} \mathrm{OH}\right): \lambda \max 382 \mathrm{~nm}$; ir (KBr): 3387, 3213, 3102, 3068, 2222 (CN), 1672, 1605, 1537, 1482, 1444, 1330, 1270, 1204, 1063, 1028, 888, 745, $691 \mathrm{~cm}^{-1}$; ${ }^{1} \mathrm{H} \mathrm{nmr}\left(\mathrm{CDCl}_{3} / \mathrm{CF}_{3} \mathrm{COOH}\right): \delta 7.20(\mathrm{~s}, 1 \mathrm{H}$, thiazol proton), 7.30-7.65 ( $\mathrm{m}, 10 \mathrm{H}$, phenyl protons), $10.35(\mathrm{~s}, 1 \mathrm{H}, \mathrm{NH}) ;{ }^{13} \mathrm{C} \mathrm{nmr}$ $\left(\mathrm{CDCl}_{3} / \mathrm{CF}_{3} \mathrm{COOH}\right): \delta \quad 104.3,108.6,109.2,117.0$ (2C), 126.6 (2C), 126.8, 127.9, 129.8 (2C), 130.0 (2C), 131.3, 139.6, 142.0, 160.3, 162.4; hrms: calcd. for $\mathrm{C}_{18} \mathrm{H}_{13} \mathrm{~N}_{5} \mathrm{OS}\left(\mathrm{M}^{+}\right) 347.0840$, found 347.0843.
$N$-(4-Phenyl-thiazol-2-yl)-2-(p-tolylhydrazono)-2-cyanoacetamide (15b).

This compound was obtained as yellow crystals, $\mathrm{mp} 230^{\circ}$; yield: $2.64 \mathrm{~g}(73 \%)$; uv/visible $\left(\mathrm{CH}_{3} \mathrm{OH}\right): ~ \lambda \max 387 \mathrm{~nm}$; ir (KBr): 3392, 3225, 3104, 3062, 2221 (CN), 1674, 1604, 1537, 1482, 1444, 1329, 1274, 1206, 1061, 1028, 887, 814, $746 \mathrm{~cm}^{-1}$; ${ }^{1} \mathrm{H} \mathrm{nmr}\left(\mathrm{CDCl}_{3} / \mathrm{CF}_{3} \mathrm{COOH}\right): \delta 2.35(\mathrm{~s}, 3 \mathrm{H}), 7.20(\psi \mathrm{~d}, 2 \mathrm{H}$, phenyl protons), 7.25 ( $\mathrm{s}, 1 \mathrm{H}$, thiazol proton), 7.45-7.65 ( $\mathrm{m}, 7 \mathrm{H}$, phenyl protons), $10.35(\mathrm{~s}, 1 \mathrm{H}, \mathrm{NH}) ;{ }^{13} \mathrm{C} \mathrm{nmr}\left(\mathrm{CDCl}_{3} /\right.$ $\left.\mathrm{CF}_{3} \mathrm{COOH}\right): \delta 20.9,103.7,108.5,109.4,116.9$ (2C), 126.6 (2C), 126.9, 129.5, 129.8 (2C), 130.6 (2C), 131.3, 137.4, 138.4, 142.0, 162.0; hrms: calcd. for $\mathrm{C}_{19} \mathrm{H}_{15} \mathrm{~N}_{5} \mathrm{OS}\left(\mathrm{M}^{+}\right) 361.0997$, found 361.0999 .
$N$-(4-Phenyl-thiazol-2-yl)-2-(p-methoxyphenylhydrazono)-2cyanoacetamide (15c).
This compound was obtained as yellow crystals, $\mathrm{mp} 251^{\circ}$; yield: $3.24 \mathrm{~g}(86 \%)$; uv/visible $\left(\mathrm{CH}_{3} \mathrm{OH}\right)$ : $\lambda$ max 399 nm ; ir (KBr): 3395, 3192, 3107, 2222 (CN),, 1670, 1611, 1537, 1482, 1441, 1328, 1274, 1251, 1206, 1165, 1059, 1025, 887, 826, $747 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H} \mathrm{nmr}$ $\left(\mathrm{CDCl}_{3} / \mathrm{CF}_{3} \mathrm{COOH}\right): \delta 3.90(\mathrm{~s}, 3 \mathrm{H}), 7.00(\psi \mathrm{~d}, 2 \mathrm{H}$, phenyl protons), 7.25 (s, 1 H , thiazol proton), $7.45-7.60(\mathrm{~m}, 7 \mathrm{H}$, phenyl protons), 10.45 (s, $1 \mathrm{H}, \mathrm{NH}$ ); ${ }^{13} \mathrm{C} \mathrm{nmr}\left(\mathrm{CDCl}_{3} / \mathrm{CF}_{3} \mathrm{COOH}\right): \delta 56.0$, 103.1, 108.5, 109.7, 115.6 (2C), 118.8 (2C), 126.6 (2C), 126.8, 129.8 (2C), 131.4, 133.9, 142.0, 158.8, 160.6, 162.4; ms: m/z (\%) 377 ( $\mathrm{M}^{+}, 100$ ), 242, 203, 176, 134, 122, 107, 77; hrms: calcd. for $\mathrm{C}_{19} \mathrm{H}_{15} \mathrm{~N}_{5} \mathrm{O}_{2} \mathrm{~S}\left(\mathrm{M}^{+}\right)$377.0946, found 377.0946 .
General Procedure for the Synthesis of N -(5-Arylazo-thiazol-2-yl)-2-arylhydrazono-2-cyanoacetamides ( $\mathbf{1 6}$ and 17).

A solution of sodium nitrite ( $0.70 \mathrm{~g}, 10.0 \mathrm{mmol}$ in 10 ml water $)$ was gradually added to a well cooled solution of the aromatic amine ( 10.0 mmol ) in concentrated $\mathrm{HCl}(3 \mathrm{ml})$. The diazonium salt solution was added with constant stirring to a cold solution of the compounds $\mathbf{8 a}$ or $\mathbf{8 b}(5.0 \mathrm{mmol})$ in pyridine ( 30 ml ). The reaction mixture was allowed to rest at $0^{\circ}$ for 2 hours, was diluted with water and then the solid was collected by filtration. The arylazo derivatives $\mathbf{1 6}$ or $\mathbf{1 7}$ thus obtained, were dried and recrystallized from an ethanol $/ N, N$-dimethylformamide mixture (1:1).
$N$-(4-Phenyl-5-phenylazo-thiazol-2-yl)-2-phenylhydrazono-2cyanoacetamide (16a).

This compound was obtained as brown crystals, mp 260-261 ${ }^{\circ}$; yield: 1.26 g ( $56 \%$ ); uv/visible $\left(\mathrm{CH}_{3} \mathrm{OH}\right)$ : $\lambda$ max 446 nm ; IR (KBr): 3401, 3228, 3062, 2222 (CN), 1675, 1604, 1554, 1521, 1480, 1446, ,1327, 1272, 1073, 891, 763, $691 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H} \mathrm{nmr}$ $\left(\mathrm{CDCl}_{3} / \mathrm{CF}_{3} \mathrm{COOH}\right): \delta \quad 7.30-7.80(\mathrm{~m}, 15 \mathrm{H}$, phenyl protons), 10.40 (s, 1H, NH); ms: m/z (\%) 451 ( $\mathrm{M}^{+}$), 416, 347, 307, 279, 242, 203 (100), 176, 134, 106, 92, 77; hrms: calcd. for $\mathrm{C}_{24} \mathrm{H}_{17} \mathrm{~N}_{7} \mathrm{OS}\left(\mathrm{M}^{+}\right) 451.1215$, found 451.1214.
$N$-[4-Phenyl-5-( $p$-methylphenylazo)-thiazol-2-yl]-2-( $p$ -methylphenylhydrazono)-2-cyanoacetamide (16b).

This compound was obtained as brown crystals, mp 241-242 ${ }^{\circ}$; yield: 1.51 g ( $63 \%$ ); uv/visible $\left(\mathrm{CH}_{3} \mathrm{OH}\right)$ : $\lambda \max 450 \mathrm{~nm}$; ir (KBr): 3392, 3225, 3033, 2921, 2218 (CN), 1677, 1603, 1520, 1481, 1445, 1327, 1277, 1211, 1073, 1030, 891, 816, 744, 693 $\mathrm{cm}^{-1} ;{ }^{1} \mathrm{H} \mathrm{nmr}\left(\mathrm{CDCl}_{3} / \mathrm{CF}_{3} \mathrm{COOH}\right): \delta 2.35(\mathrm{~s}, 3 \mathrm{H}), 2.40(\mathrm{~s}, 3 \mathrm{H})$, 7.20-7.85 (m, 13H, phenyl protons), 10.10 (s, $1 \mathrm{H}, \mathrm{NH}$ ); ms: m/z (\%) 479( $\left.{ }^{+}, 100\right), 446,361,321,293,266,203$ (100), 176, 134, 106, 92, 77; hrms: calcd. for $\mathrm{C}_{26} \mathrm{H}_{21} \mathrm{~N}_{7} \mathrm{OS}\left(\mathrm{M}^{+}\right) 479.1528$, found 479.1531.
$N$-[4-Phenyl-5-( $p$-methoxyphenylazo)-thiazol-2-yl]-2-( $p$ -methoxyphenylhydrazono)-2-cyanoacetamide (16c).

This compound was obtained as red-brown crystals, mp $248^{\circ}$; yield: 1.38 g ( $54 \%$ ); uv/visible $\left(\mathrm{CH}_{3} \mathrm{OH}\right)$ : $\lambda$ max 468 nm ; ir (KBr): 3402, 3202, 3067, 2938, 2217 (CN), 1676, 1599, 1578, 1521, 1479, 1327, 1279, 1250, 1081, 1030, 892, $827 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ nmr (DMSO- $\mathrm{d}_{6}$ ): $\delta 3.80(\mathrm{~s}, 3 \mathrm{H}), 3.85(\mathrm{~s}, 3 \mathrm{H}), 6.95(\psi \mathrm{~d}, 2 \mathrm{H}$, phenyl protons), 7.05 ( $\psi \mathrm{d}, 2 \mathrm{H}$, phenyl protons), $7.35-7.50$ ( m , 3 H , phenyl protons), $7.75(\mathrm{~m}, 4 \mathrm{H}$, phenyl protons), $8.25(\psi \mathrm{~d}, 2 \mathrm{H}$, phenyl protons), $12.10(\mathrm{~s}, 1 \mathrm{H}, \mathrm{NH}), 12.45(\mathrm{~s}, 1 \mathrm{H}, \mathrm{NH}) ;{ }^{13} \mathrm{C} \mathrm{nmr}$ (DMSO- $\mathrm{d}_{6}$ ): $\delta 53.6,53.8,102.9,109.2,112.5$ (2C), 112.9 (2C), 116.8 (2C), 122.5 (2C), 126.5 (2C), 127.2, 128.0 (2C), 132.0,
133.7, 142.7, 144.8, 147.4, 155.5, 156.4, 158.9, 159.,8; ms: m/z (\%) $511\left(\mathrm{M}^{+}\right), 481,416,377,335,310(100), 279,229,214,176$, 133, 107, 77.
$N$-[4-(4-Antipyrinyl)-5-phenylazo-thiazol-2-yl)-2-(phenylhydra-zono)-2-cyanoacetamide (17a).

This compound was obtained as orange crystals, mp 276-277 ${ }^{\circ}$; yield: 1.77 g ( $63 \%$ ); uv/visible $\left(\mathrm{CH}_{3} \mathrm{OH}\right)$ : $\lambda \max 456 \mathrm{~nm}$; ir (KBr): 3393, 2925, 2211 (CN), 1687, 1646, 1608, 1515, 1484, 1346, 1271, 1206, $1143 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H} \mathrm{nmr}\left(\mathrm{CF}_{3} \mathrm{COOD}\right): \delta 2.90$ (s, $3 \mathrm{H}), 3.65(\mathrm{~s}, 3 \mathrm{H}), 7.25-7.80\left(\mathrm{~m}, 15 \mathrm{H}\right.$, phenyl protons); ${ }^{13} \mathrm{C} \mathrm{nmr}$ ( $\mathrm{CF}_{3} \mathrm{COOD}$ ): $\delta 13.0,33.7,97.5,104.4,109.2,117.1$ (2C), 123.4 (2C), 128.2, 128.6 (2C), 129.5, 130.1 (2C), 130.1 (2C), 131.1 (2C), 133.1, 133.8, 136.4, 139.5, 143.4, 148.9, 150.9, 157.8, $160.1,160.5 ; \mathrm{ms}: \mathrm{m} / \mathrm{z}(\%) 561\left(\mathrm{M}^{+}\right), 534,506,457,430,362$, 312, 286, 269, 234, 221, 169, 161, 145 (100), 91, 77, 56; hrms: calcd. for $\mathrm{C}_{29} \mathrm{H}_{23} \mathrm{~N}_{9} \mathrm{O}_{2} \mathrm{~S}\left(\mathrm{M}^{+}\right) 561.1695$, found 561.1693.
$N$-[4-(4-Antipyrinyl)-5-( $p$-methyl-phenylazo)-thiazol-2-yl]-2-( $p$ -methyl-phenylhydrazono)-2-cyanoacetamide (17b).

This compound was obtained as orange crystals, mp 283-284 ${ }^{\circ}$; yield: 2.00 g ( $68 \%$ ); uv/visible $\left(\mathrm{CH}_{3} \mathrm{OH}\right)$ : $\lambda \max 461 \mathrm{~nm}$; ir (KBr): 3393, 2918, 2210 (CN), 1682, 1651, 1610, 1515, 1484, 1413, 1273, 1212, $1070 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H} \mathrm{nmr}\left(\mathrm{CDCl}_{3} / \mathrm{CF}_{3} \mathrm{COOH}\right): \delta$ $2.35(\mathrm{~s}, 3 \mathrm{H}), 2.40(\mathrm{~s}, 3 \mathrm{H}), 2.85(\mathrm{~s}, 3 \mathrm{H}), 3.60(\mathrm{~s}, 3 \mathrm{H}), 7.15-7.75$ ( $\mathrm{m}, 13 \mathrm{H}$, phenyl protons), $10.10(\mathrm{~s}, 1 \mathrm{H}, \mathrm{NH}) ;{ }^{13} \mathrm{C} \mathrm{nmr}\left(\mathrm{CDCl}_{3} /\right.$ $\left.\mathrm{CF}_{3} \mathrm{COOH}\right): \delta 12.9,20.8,21.6,33.7,97.7,104.5,109.4,116.6$ (2C), 123.3 (2C), 128.5 (2C), 129.6, 130.5 (2C); 130.6 (2C), 130.8 (2C), 132.6, 136.7, 137.3, 137.9, 143.1, 144.9, 148.8, 157.8, 159.4, 159.6, 160.0.
$N$-[4-(4-Antipyrinyl)-5-(p-methoxyphenylazo)-thiazol-2-yl]-2( $p$-methoxy-phenylhydrazono)-2-cyanoacetamide (17c).

This compound was obtained as orange crystals, $\mathrm{mp} 277-278^{\circ}$; yield: 2.27 g ( $73 \%$ ); uv/visible $\left(\mathrm{CH}_{3} \mathrm{OH}\right)$ : $\lambda$ max 471 nm ; ir (KBr): 3392, 2939, 2210 (CN), 1678, 1646, 1597, 1510, 1478, 1436, 1246, 1146, $1068 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H} \mathrm{nmr}\left(\mathrm{CDCl}_{3} / \mathrm{CF}_{3} \mathrm{COOH}\right): \delta$ $2.85(\mathrm{~s}, 3 \mathrm{H}), 3.60(\mathrm{~s}, 3 \mathrm{H}), 3.85(\mathrm{~s}, 3 \mathrm{H}), 3.90(\mathrm{~s}, 3 \mathrm{H}), 6.90-7.70$ (m, 13 H , phenyl protons); ${ }^{13} \mathrm{C} \mathrm{nmr}\left(\mathrm{CDCl}_{3} / \mathrm{CF}_{3} \mathrm{COOH}\right): \delta 12.8$, 33.7, 55.8, 55.9, 97.7, 103.6, 109.6, 115.5 (4C), 118.5 (2C), 126.1 (2C), 128.5 (2C), 129.5, 130.9 (2C), 132.8, 133.7, 134.6, 143.3, 145.0, 148.5, 157.7, 158.8, 159.7, 160.0, 164.1.

General Procedure for the Synthesis of 3-(5-Arylazo-thiazol-2-yl)-rhodanines ( 18 and 19).

A solution of sodium nitrite ( $0.35 \mathrm{~g}, 5.0 \mathrm{mmol}$ in 5 ml water) was gradually added to a well cooled and stirred suspension of the aromatic amine $(5.0 \mathrm{mmol})$ in concentrated $\mathrm{HCl}(1.5 \mathrm{ml})$. The diazonium salt solution was added with constant stirring to a cold solution $\left(0-5^{\circ}\right)$ of compound $\mathbf{1 3}(5.0 \mathrm{mmol})$ in pyridine ( 30 $\mathrm{ml})$. The reaction mixture was allowed to rest at $0^{\circ}$ for 2 hours, it was diluted with water and then the solid was collected by filtration. The solid arylazo derivatives $\mathbf{1 8}$ or $\mathbf{1 9}$ thus obtained, were dried and recrystallized from DMF with addition of water.

## 3-(4-Phenyl-5-phenylazo-thiazol-2-yl)-rhodanine (18a).

This compound was obtained as brown crystals, $\mathrm{mp} 212-213^{\circ}$; yield: 1.33 g ( $67 \%$ ); uv/visible $\left(\mathrm{CH}_{3} \mathrm{OH}\right): ~ \lambda \max 433 \mathrm{~nm}$; ir (KBr): 2918, 1694, 1592, 1567, 1516, 1458,1415, 1393, 1283,

1242, 1016, $913 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H} \mathrm{nmr}\left(\mathrm{CDCl}_{3} / \mathrm{CF}_{3} \mathrm{COOD}\right): \delta 4.25$ (s, $2 \mathrm{H}), 7.45-7.60(\mathrm{~m}, 6 \mathrm{H}$, phenyl protons), $7.80(\psi \mathrm{~d}, 2 \mathrm{H}$, phenyl protons), 7.90 ( $\psi \mathrm{d}, 2 \mathrm{H}$, phenyl protons); ${ }^{13} \mathrm{C} \mathrm{nmr}\left(\mathrm{CDCl}_{3} /\right.$ $\mathrm{CF}_{3} \mathrm{COOD}$ ): $\delta \quad 37.4,123.6$ (2C), 126.3, 129.5 (2C), 129.7 (4C), 132.1, 133.2, 138.8, 140.3, 151.3, 162.6, 173.8, 195.5; ms: m/z (\%) $396\left(\mathrm{M}^{+}\right), 321$ (100), 279, 173, 133, 102, 89, 77; hrms: calcd. for $\mathrm{C}_{18} \mathrm{H}_{12} \mathrm{~N}_{4} \mathrm{OS}_{3}\left(\mathrm{M}^{+}\right)$396.0173, found 396.0176 .

## 3-[4-Phenyl-5-( $p$-tolylazo)-thiazol-2-yl]-rhodanine (18b).

This compound was obtained as red-brown crystals, mp 222$223^{\circ}$; yield: 1.46 g ( $71 \%$ ); uv/visible $\left(\mathrm{CH}_{3} \mathrm{OH}\right)$ : $\lambda$ max 430 nm ; ir (KBr): 2920, 1700, 1598, 1563, 1516, 1451, 1411, 1386, 1282, 1240, 1208, $1014 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H} \mathrm{nmr}\left(\mathrm{CDCl}_{3} / \mathrm{CF}_{3} \mathrm{COOD}\right): \delta 2.40(\mathrm{~s}$, $3 \mathrm{H}), 4.25(\mathrm{~s}, 2 \mathrm{H}), 7.35$ ( $\psi \mathrm{d}, 2 \mathrm{H}$, phenyl protons), $7.55(\mathrm{~s}, 3 \mathrm{H}$, phenyl protons), 7.75 ( $\psi \mathrm{d}, 2 \mathrm{H}$, phenyl protons), $7.90(\psi \mathrm{~d}, 2 \mathrm{H}$, phenyl protons); ${ }^{13} \mathrm{C} \mathrm{nmr}\left(\mathrm{CDCl}_{3} / \mathrm{CF}_{3} \mathrm{COOD}\right): \delta 21.5,37.4$, 123.6 (2C), 126.5, 129.6 (2C), 129.7 (2C), 130.5 (2C), 132.2, 140.7, 144.7, 145.0, 149.1, 162.7, 173.9, 195.4; ms: m/z (\%) 410 $\left(\mathrm{M}^{+}\right), 393,335$ (100), 293, 173, 133, 102, 91, 65.

## 3-[4-(4-Antipyrinyl)-5-phenylazo-thiazol-2-yl]-rhodanine (19a).

This compound was obtained as violet-brown crystals, mp $160-161^{\circ}$; yield: $1.85 \mathrm{~g}(73 \%)$; uv/visible $\left(\mathrm{CH}_{3} \mathrm{OH}\right)$ : $\lambda \max 440$ nm; ir (KBr): 2970, 1727, 1617, 1578, 1478, 1419, 1284, 1247, 1191, 1146, 1048, $992 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H} \mathrm{nmr}\left(\mathrm{CDCl}_{3} / \mathrm{CF}_{3} \mathrm{COOD}\right): \delta 2.85$ $(\mathrm{s}, 3 \mathrm{H}), 3.60(\mathrm{~s}, 3 \mathrm{H}), 4.20(\mathrm{~s}, 2 \mathrm{H}), 7.30-7.70(\mathrm{~m}, 10 \mathrm{H}$, phenyl protons); ${ }^{13} \mathrm{C} \mathrm{nmr}\left(\mathrm{CDCl}_{3} / \mathrm{CF}_{3} \mathrm{COOD}\right): \delta 12.9,33.7,37.3,98.7$, 122.8 (2C), 128.5 (2C), 129.4, 129.8 (2C), 130.8 (2C), 132.7 (2C), 139.9, 140.8, 149.3, 150.5, 156.8, 161.6, 172.5, 194.3; ms: $\mathrm{m} / \mathrm{z}(\%) 506\left(\mathrm{M}^{+}\right), 430(100), 390,362,338,298,282,242,240$, 169, 119, 93, 77, 56.
3-[4-(4-Antipyrinyl)-5-(p-tolylazo)-thiazol-2-yl]-rhodanine (19b).
This compound was obtained as violet-brown crystals, mp $175-176^{\circ}$; yield: $1.51 \mathrm{~g}(58 \%)$; uv/visible $\left(\mathrm{CH}_{3} \mathrm{OH}\right)$ : $\lambda \max 443$ nm; ir (KBr): 2925, 1723, 1616, 1583, 1476, 1420, 1328, 1281, 1247, 1144, 988, $751 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H} \mathrm{nmr}\left(\mathrm{CDCl}_{3} / \mathrm{CF}_{3} \mathrm{COOD}\right): \delta 2.40$ (s, 3H), $2.9(\mathrm{~s}, 3 \mathrm{H}), 3.7(\mathrm{~s}, 3 \mathrm{H}), 4.3(\mathrm{~s}, 2 \mathrm{H}), 7.3-7.8(\mathrm{~m}, 9 \mathrm{H}$, phenyl protons); ${ }^{13} \mathrm{C} \mathrm{nmr}\left(\mathrm{CDCl}_{3} / \mathrm{CF}_{3} \mathrm{COOD}\right): \delta 12.9,21.6$, 33.7, 37.2, 98.2, 123.3 (2C), 128.5 (2C), 129.3, 130.8 (2C), 131.0 (2C), 133.0 (2C), 138.1, 140.1, 148.6, 148.8, 157.1, 162.1, 174.0, 193.4; ms: m/z (\%) $520\left(\mathrm{M}^{+}\right), 466,434,404,393,328,317,286$ (100), 213, 188, 149, 124, 106, 77, 56.

General Procedure for the Synthesis of 2-Cyano-3-( $p$-tolyl)- $N$-(5-arylazo-thiazol-2-yl)-2-propeneamides ( $\mathbf{2 0}$ and 21).

A solution of sodium nitrite ( $0.35 \mathrm{~g}, 5.0 \mathrm{mmol}$ in 5 ml water ) was gradually added to a well cooled solution of the aromatic amine ( 5.0 mmol ) in concentrated $\mathrm{HCl}(1.5 \mathrm{ml})$. The diazonium salt solution was added with constant stirring to a cold solution of compound $9(5.0 \mathrm{mmol})$ in pyridine ( 50 ml ). The reaction mixture was stirred at $0^{\circ}$ for 3 hours and the solid product was collected by filtration. The arylazo derivatives $\mathbf{2 0}$ or $\mathbf{2 1}$ were dried and recrystallized from DMF.

2-Cyano-3-( $p$-tolyl)- N -(4-phenyl-5-phenylazo-thiazol-2-yl)-2propeneamide (20a).

This compound was obtained as orange crystals, mp 241-242 ${ }^{\circ}$; yield: 2.04 g ( $91 \%$ ); uv/visible $\left(\mathrm{CH}_{3} \mathrm{OH}\right)$ : $\lambda \max 424 \mathrm{~nm}$; ir (KBr): 3058, 2899, 2227(CN), 1646, 1605, 1580, 1479, 1443,

1381, 1322, 1302, $1282 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H} \mathrm{nmr}\left(\mathrm{CF}_{3} \mathrm{COOD}\right): \delta 2.45$ (s, $3 \mathrm{H}), 7.35(\psi \mathrm{~d}, 2 \mathrm{H}$, phenyl protons), $7.50-8.00(\mathrm{~m}, 12 \mathrm{H}$, phenyl protons), $8.55(\mathrm{~s}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C} \mathrm{nmr}\left(\mathrm{CF}_{3} \mathrm{COOD}\right): \delta 22.0,99.6,123.8$ (2C), 126.4, 128.3, 129.5 (4C), 129.6 (2C), 130.6 (2C), 132.0, 132.3 (2C), 133.3, 140.6, 142.8, 147.7, 151.6, 158.4, 161.5, 161.7 (together with a signal of the solvent), 162.7; hrms: calcd. for $\mathrm{C}_{26} \mathrm{H}_{19} \mathrm{~N}_{5} \mathrm{OS}\left(\mathrm{M}^{+}\right) 449.1310$, found 449.1309.

2-Cyano-3-( $p$-tolyl)- $N$-[4-phenyl-5-( $p$-tolylazo)-thiazol-2-yl]-2propeneamide (20b).

This compound was obtained as orange crystals, mp 263$264^{\circ}$; yield: 2.04 g ( $88 \%$ ); uv/visible $\left(\mathrm{CH}_{3} \mathrm{OH}\right)$ : $\lambda$ max 429 nm ; ir ( KBr ): 3063, 2919, 2227 (CN), 1646, 1604, 1561, 1484, 1444, 1380, 1321, 1302, $1282 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H} \mathrm{nmr}\left(\mathrm{CF}_{3} \mathrm{COOD}\right): \delta 2.45$ (s, 6 H ), 7.30-7.40 (m, 4H, phenyl protons), 7.60 (br s, 3 H , phenyl protons), $7.80(\psi \mathrm{~d}, 2 \mathrm{H}$, phenyl protons), $7.95(\psi \mathrm{~d}, 4 \mathrm{H}$, phenyl protons), $8.55(\mathrm{~s}, 1 \mathrm{H})$.

2-Cyano-3-(p-tolyl)- N -[4-(4-antipyrinyl)-5-phenylazo-thiazol-2-yl)-2-propeneamide (21a).

This compound was obtained as orange-yellow crystals, mp 261-262 ${ }^{\circ}$; yield: $2.12 \mathrm{~g}(76 \%)$; uv/visible $\left(\mathrm{CH}_{3} \mathrm{OH}\right): \lambda_{\text {max }}=455$ nm ; ir (KBr): 3056, 2953, 2218 (CN), 1669, 1618, 1586, 1538, 1495, 1424, 1370, 1320, $1301 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H} \mathrm{nmr}\left(\mathrm{CF}_{3} \mathrm{COOD}\right): \delta 2.50$ $(\mathrm{s}, 3 \mathrm{H}), 3.05(\mathrm{~s}, 3 \mathrm{H}), 3.75(\mathrm{~s}, 3 \mathrm{H}), 7.40-7.85(\mathrm{~m}, 12 \mathrm{H}$, phenyl protons), $8.00(\psi \mathrm{~d}, 2 \mathrm{H}$, phenyl protons), $8.70(\mathrm{~s}, 1 \mathrm{H}) ; \mathrm{ms}: \mathrm{m} / \mathrm{z}(\%)$ $559\left(\mathrm{M}^{+}\right), 531,480,466,455(100), 388,343,313,286,243,195$, 181, 170, 115, 93, 77, 56.

2-Cyano-3-( $p$-tolyl)- $N$-[4-(4-antipyrinyl)-5-( $p$-tolylazo)-thiazol-2-yl)-2-propeneamide (21b).

This compound was obtained as orange crystals, mp 237-238 ${ }^{\circ}$, yield: 2.35 g ( $82 \%$ ); uv/visible $\left(\mathrm{CH}_{3} \mathrm{OH}\right)$ : $\lambda \max 460 \mathrm{~nm}$; ir (KBr): 3046, 2922, 2214 (CN), 1650, 1602, 1589, 1494, 1421, 1374, 1299, 1205, $1181 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H} \mathrm{nmr}\left(\mathrm{CF}_{3} \mathrm{COOD}\right): \delta 2.45(\mathrm{~s}$, $3 \mathrm{H}), 2.50(\mathrm{~s}, 3 \mathrm{H}), 3.00(\mathrm{~s}, 3 \mathrm{H}), 3.70(\mathrm{~s}, 3 \mathrm{H}), 7.30-7.75(\mathrm{~m}, 11 \mathrm{H}$, phenyl protons), $8.00\left(\psi \mathrm{~d}, 2 \mathrm{H}\right.$, phenyl protons), $8.65(\mathrm{~s}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}$ $\mathrm{nmr}\left(\mathrm{CF}_{3} \mathrm{COOD}\right): \delta 13.1,21.5,21.8,33.7,97.9,98.9,115.5$, 123.6 (2C), 128.5 (2C), 129.4, 130.8 (2C), 130.9 (2C), 131.2 (2C), 132.4 (2C), 133.2, 139.0, 144.0, 145.9, 148.4, 148.7, 149.2, 157.0, 159.7, 160.1, 160.5, 165.9; ms: m/z (\%) $573\left(\mathrm{M}^{+}\right), 545,480,466$, 455 (100), 404, 388, 343, 313, 286, 243, 195, 170, 115, 91.

## REFERENCES AND NOTES

[1] J. V. Metzger, in Comprehensive Heterocyclic Chemistry, Vol 6, K. T. Potts, ed, Pergamon Press, Oxford, 1984, pp 235-331; R. Barone, M. Chanon and R. Gallo, in Thiazole and its Derivatives, Vol. 34/2, J. V. Metzger, ed, Wiley, New York, 1979, p. 9ff.; L. Forlani, Targets in Heterocyclic Systems, 1, 75 (1997).
[2] N. Amishiro, S. Nagamuro, E. Kobayashi, K. Gomi and H. Saito, J. Med. Chem., 42, 669 (1999).
[3] K. Hattori, N. Fujii, A. Tanaka and F. Takamura, WO Patent 99/21843 to Fujisawa Pharmaceutical Co., LTD; Chem. Abstr., 130, 325140s (1999).
[4] M. Cheung, K. C. Glennon, K. E. Lackey and M. R. Pell, WO Patent 99/21859 to Glaxo Group LTD ; Chem. Abstr., 130, 325139y (1999).
[5] K. Yamaguchi, M. Yada, T. Tsuji, Y. Hatanaka, K. Goda and T. Kobori, Bioorg. \& Med. Chem. Lett., 9, 957 (1999).

Table 1
Elemental Analyses of the New Compounds

| Compound | Formula | C: Calc. (Found) | H: Calc. (Found) | N: Calc. (Found) |
| :---: | :---: | :---: | :---: | :---: |
| 5 | $\mathrm{C}_{16} \mathrm{H}_{16} \mathrm{~N}_{4} \mathrm{O}_{2} \mathrm{~S}$ | 58.53 (58.67) | 4.88 (4.95) | 17.07 (16.88) |
| 6 | $\mathrm{C}_{16} \mathrm{H}_{15} \mathrm{ClN}_{4} \mathrm{O}_{2} \mathrm{~S}$ | 52.96 (52.89) | 4.14 (4.10) | 15.45 (15.40) |
| 7 a | $\mathrm{C}_{23} \mathrm{H}_{19} \mathrm{~N}_{5} \mathrm{O}_{2} \mathrm{~S}_{3}$ | 55.98 (55.87) | 3.85 (3.88) | 14.20 (14.25) |
| 7b | $\mathrm{C}_{23} \mathrm{H}_{19} \mathrm{~N}_{5} \mathrm{O}_{3} \mathrm{~S}_{2}$ | 57.86 (57.91) | 3.98 (3.82) | 14.67 (14.75) |
| 7c | $\mathrm{C}_{23} \mathrm{H}_{20} \mathrm{~N}_{6} \mathrm{O}_{2} \mathrm{~S}_{2}$ | 57.98 (57.86) | 4.20 (4.27) | 17.65 (17.56) |
| 9 a | $\mathrm{C}_{20} \mathrm{H}_{15} \mathrm{~N}_{3} \mathrm{OS}$ | 69.56 (69.43) | 4.35 (4.22) | 12.17 (12.04) |
| 9b | $\mathrm{C}_{25} \mathrm{H}_{21} \mathrm{~N}_{5} \mathrm{O}_{2} \mathrm{~S}$ | 65.93 (65.82) | 4.61 (4.53) | 15.38 (15.21) |
| 11b | $\mathrm{C}_{16} \mathrm{H}_{16} \mathrm{~N}_{4} \mathrm{OS}_{3}$ | 51.06 (51.11) | 4.25 (4.22) | 14.89 (14.83) |
| 12b | $\mathrm{C}_{17} \mathrm{H}_{18} \mathrm{~N}_{4} \mathrm{OS}_{3}$ | 52.31 (52.39) | 4.61 (4.47) | 14.36 (14.22) |
| 13a | $\mathrm{C}_{12} \mathrm{H}_{8} \mathrm{~N}_{2} \mathrm{OS}_{3}$ | 49.31 (49.43) | 2.74 (2.87) | 9.59 (9.73) |
| 13b | $\mathrm{C}_{17} \mathrm{H}_{14} \mathrm{~N}_{4} \mathrm{O}_{2} \mathrm{~S}_{3}$ | 50.74 (50.56) | 3.48 (3.32) | 13.93 (14.05) |
| 14b | $\mathrm{C}_{21} \mathrm{H}_{17} \mathrm{~N}_{5} \mathrm{OS}_{2}$ | 60.14 (60.03) | 4.06 (4.17) | 16.70 (16.54) |
| 15a | $\mathrm{C}_{18} \mathrm{H}_{13} \mathrm{~N}_{5} \mathrm{OS}$ | 62.24 (62.22) | 3.74 (3.68) | 20.17 (20.21) |
| 15b | $\mathrm{C}_{19} \mathrm{H}_{15} \mathrm{~N}_{5} \mathrm{OS}$ | 63.16 (63.24) | 4.15 (4.19) | 19.39 (19.43) |
| 15c | $\mathrm{C}_{19} \mathrm{H}_{15} \mathrm{~N}_{5} \mathrm{O}_{2} \mathrm{~S}$ | 60.48 (60.44) | 3.98 (4.05) | 18.57 (18.63) |
| 16a | $\mathrm{C}_{24} \mathrm{H}_{17} \mathrm{~N}_{7} \mathrm{OS}$ | 63.86 (64.11) | 3.77 (3.91) | 21.73 (21.55) |
| 16b | $\mathrm{C}_{26} \mathrm{H}_{21} \mathrm{~N}_{7} \mathrm{OS}$ | 65.13 (65.28) | 4.38 (4.21) | 20.46 (20.58) |
| 16c | $\mathrm{C}_{26} \mathrm{H}_{21} \mathrm{~N}_{7} \mathrm{O}_{3} \mathrm{~S}$ | 61.05 (60.92) | 4.11 (4.23) | 19.18 (19.33) |
| 17a | $\mathrm{C}_{29} \mathrm{H}_{23} \mathrm{~N}_{9} \mathrm{O}_{2} \mathrm{~S}$ | 62.03 (62.25) | 4.10 (4.24) | 22.46 (22.63) |
| 17b | $\mathrm{C}_{31} \mathrm{H}_{27} \mathrm{~N}_{9} \mathrm{O}_{2} \mathrm{~S}$ | 63.16 (63.41) | 4.58 (4.80) | 21.39 (21.23) |
| 17c | $\mathrm{C}_{31} \mathrm{H}_{27} \mathrm{~N}_{9} \mathrm{O}_{4} \mathrm{~S}$ | 59.90 (60.13) | 4.35 (4.49) | 20.29 (20.42) |
| 18a | $\mathrm{C}_{18} \mathrm{H}_{12} \mathrm{~N}_{4} \mathrm{OS}_{3}$ | 54.54 (54.38) | 3.03 (3.12) | 14.14 (14.03) |
| 18b | $\mathrm{C}_{19} \mathrm{H}_{14} \mathrm{~N}_{4} \mathrm{OS}_{3}$ | 55.61 (55.66) | 3.41 (3.52) | 13.66 (13.43) |
| 19a | $\mathrm{C}_{23} \mathrm{H}_{18} \mathrm{~N}_{6} \mathrm{O}_{2} \mathrm{~S}_{3}$ | 54.54 (54.45) | 3.56 (3.43) | 16.60 (16.51) |
| 19b | $\mathrm{C}_{24} \mathrm{H}_{20} \mathrm{~N}_{6} \mathrm{O}_{2} \mathrm{~S}_{3}$ | 55.38 (55.49) | 3.84 (3.93) | 16.15 (15.88) |
| 20a | $\mathrm{C}_{26} \mathrm{H}_{19} \mathrm{~N}_{5} \mathrm{OS}$ | 69.49 (69.44) | 4.23 (4.29) | 15.59 (15.64) |
| 20 b | $\mathrm{C}_{27} \mathrm{H}_{21} \mathrm{~N}_{5} \mathrm{OS}$ | 69.98 (69.93) | 4.53 (4.43) | 15.12 (15.01) |
| 21a | $\mathrm{C}_{31} \mathrm{H}_{25} \mathrm{~N}_{7} \mathrm{O}_{2} \mathrm{~S}$ | 66.55 (66.59) | 4.47 (4.53) | 17.53 (17.42) |
| 21b | $\mathrm{C}_{32} \mathrm{H}_{27} \mathrm{~N}_{7} \mathrm{O}_{2} \mathrm{~S}$ | 67.01 (66.88) | 4.71 (4.82) | 17.10 (17.21) |

[6] I. Hayakawa, Y. Tanaka and Y. Nagata, Japan Kokai, 77, 83, 588 (1976); Chem. Abstr., 88, 37785c (1978).
[7] T. Y. Shen, R. L. Olark, A. A. Pessolano, B. E. Witzel, E. Bruce and T. J. Lonza, U. S. Patent 4038396 (1975); Chem. Abstr., 78, 152185y (1977).
[8] H. Beyer and D. Stehwien, Arch. Pharm., 286, 13 (1953).
[9] G. Kaupp, J. Schmeyers and J. Boy, J. Prakt. Chem., 342, 269 (2000).
[10] G. Kaupp, in Comprehensive Supramolcular Chemistry, Vol. 8, J. L. Atwood, J. E. D. Davies, D. D. MacNicol, F. Voegtle, J.-M. Lehn and J. A. Ripmeester, ed, Elsevier, Oxford, 1996, pp. 381-423 and 21 color plates.
[11] B. Dash, P. K. Mahapatra, D. Panda and J. M. Pattnaik, J. Ind. Chem. Soc., 61, 1061 (1984).
[12] P. N. Bhargava and S. C. Sharma, Bull. Chem. Soc. Japan, 38, 909 (1965).
[13] A. M. Mahmoud, H. A. El-Sherief and A. E. Abdel-Raman, Eur. J. Med. Chem. Chim. Ther., 16, 383 (1981).
[14] V. Ariesan, Z. Cojocaru, M. Moga-Iuga, D. Ghiran, E.

Chindris and L. Satfa, Farmacia (Bucharest), 19, 65 (1971).
[15] K. M. Dawood, A. M. Farag, E. A. Ragab and Z. E. Kandeel, J. Chem. Res. Miniprint, 622 (2000).
[16] R. H. Khan and R. C. Rastogi, Ind. J. Chem. Soc., Sect. B, 32, 595 (1993).
[17] R. H. Khan, R. C. Rastogi and C. Ramesh, J. Chem. Res. Synopses, 160 (1991).
[18] J. Garin, E. Melendez, F. L. Merchan, D. Ortiz and T. Tejero, Synthesis, 368 (1987).
[19] D. Krishnaswamy, V. V. Govande, V. K. Gumaste, B. M. Bhawal and A. R. A. S. Deshmukh, Tetrahedron, 58, 2215 (2002).
[20] J. Garin, E. Melendez, F. L. Merchan and T. Tejero, Synthesis, 520 (1984).
[21] J. Garin, E. Melendez, F. L. Merchan and T. Tejero, Synthesis, 867 (1985).
[22] K. Bogdanowicz-Szwed and M. Krasodomska, Monatsh. Chem., 127, 1273 (1996).
[23] H. Zollinger, Diazo Chemistry, VCH, Weinheim, 1994.

